


Name:			
Enrolment No:			
UPES End Semester Examination, May 2023			
Course: Physics Program: B. Tech FSE Course Code: PHYS 1002		Semester: II Time : 03 hrs. Max. Marks: 100	
Instructions: Use of scientific calculator is allowed			
SECTION A (5Qx4M=20Marks)			
S. No.		Marks	CO
Q1	Differentiate between Fresnel and Fraunhofer diffraction	4	CO1
Q2	Prove that phase velocity exceeds the velocity of light.	4	CO3
Q3	A signal of power $5\mu\text{W}$ exists just inside the entrance of 0.1 km long fibre. Calculate the attenuation coefficient of the fibre if the power inside the fibre be $1\mu\text{W}$.	4	CO1
Q4	A particle limited to the x-axis has the wave function $\Psi = ax$ between $x = 0$ and $x = 1$; $\Psi = 0$ elsewhere. Find: (i) the probability that the particle can be found between $x = 0.45$ and $x = 0.55$. (ii) the expectation value $\langle x \rangle$ of the particle's position.	4	CO3
Q5	Distinguish between primitive cell and non-primitive cell.	4	CO4
SECTION B (4Qx10M= 40 Marks)			
Q6	Explain the hysteresis curve in ferromagnetic materials.	10	CO2
Q7	An electron is trapped in a 1D infinitely deep potential well of width $L = 10^{-9}$ m. Calculate the wavelength of photon emitted from the transition $E_4 \rightarrow E_3$.	10	CO3
Q8	Deduce the relationship between Einstein's coefficients. OR Show that plane and circularly polarized lights are the special cases of an elliptically polarized light.	10	CO1
Q9	If the earth receives $2\text{ cal min}^{-1}\text{cm}^{-2}$ solar energy, what would be the amplitudes of electric and magnetic fields of radiation	10	CO2
SECTION-C (2Qx20M=40 Marks)			
Q10	a) Derive the mathematical expression for Compton shift. (10)	20	CO3

	<p>b) What is Heisenberg's uncertainty principle? Explain why electrons cannot reside inside the nucleus? (10)</p> <p style="text-align: center;">OR</p> <p>a) What is pair production? Explain why it cannot take place in an empty space. (10)</p> <p>b) Develop the time dependent Schrodinger wave equation for a quantum particle starting with simple wave equation. (10)</p>		
Q11	<p>a) Define photovoltaic effect. Explain the construction and working of a solar cell with its I-V characteristics. (10)</p> <p>b) What are Miller indices? Draw the planes (111) and (234). (10)</p>	20	CO4

LIST OF IMPORTANT CONSTANTS

<p>Planck's constant, $h = 6.6 \times 10^{-34}$ J.s</p> <p>Boltzmann's constant, $k = 1.38 \times 10^{-23}$ J/K</p> <p>Mass of electron, $m_e = 9.1 \times 10^{-31}$ Kg</p> <p>Mass of proton, $m_p = 1.67 \times 10^{-27}$ Kg</p> <p>Velocity of light, $c = 3 \times 10^8$ m/s</p> <p>Rydberg Constant, $R = 1.097 \times 10^7$ m⁻¹</p> <p>Avogadro's number = 6.023×10^{23}</p> <p>Permeability of free space, $\mu_0 = 4\pi \times 10^{-7}$ Henry/m</p> <p>Permittivity of free space, $\epsilon_0 = 8.85 \times 10^{-12}$ F/m</p> <p>Impedance of em wave in free space $Z_0 = 377$ Ohm</p>
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