


Name:			
Enrolment No:			
UNIVERSITY OF PETROLEUM AND ENERGY STUDIES End Semester Examination, May 2022			
Course: Engineering Physics Program: B.Tech. CS (All Batches) Course Code: PHYS 1023		Semester: II Time : 03 hrs. Max. Marks: 100	
Instructions: <ul style="list-style-type: none"> • There are 3 Sections such as Section A, B & C. • Section A is compulsory, however, Section B & Section C have internal choices. • Scientific calculator is allowed. 			
SECTION A <i>(5 Q × 4 M = 20 Marks)</i>			
S. No.		Marks	CO
Q 1.	Distinguish ordinary Photography and Holography pointwise.	4	CO1
Q 2.	Prove that $\vec{E} = -\vec{\nabla} V$ where \vec{E} is electric field intensity and V is electric potential and $\vec{\nabla}$ is the gradient of V .	4	CO2
Q 3.	Define magnetic flux density and explain that an isolated magnetic charge does not exist.	4	CO3
Q 4.	Calculate the maximum percentage change in wavelength due to Compton scattering for incident photons of wavelengths 1 \AA and 10 \AA .	4	CO4
Q 5.	State the various applications of nanomaterials.	4	CO5
SECTION B <i>(4 Q × 10 M = 40 Marks)</i>			
Q 6.	Explain the construction and working of a Ruby laser using suitable diagrams.	10	CO1
Q 7.	Derive the expression for boundary conditions between two different dielectric materials along with schematic diagrams.	10	CO2
Q 8.	a) Given the magnetic flux density $\vec{B} = \frac{\mu_0}{2} \hat{a}_\phi \text{ Wb/m}$, calculate the total magnetic flux crossing the surface $\phi = \frac{\pi}{2}$, $1 < \rho < 3$ meters and $0 < z < 3$ meters	5	CO3
	b) Distinguish bits and qubits in quantum computing and define quantum confinement.	5	CO5
Q 9.	The work function for lithium is $4.6 \times 10^{-19} \text{ J}$.	10	CO4

	<p>(a) Calculate the lowest frequency of light that will cause photoelectric emission.</p> <p>(b) What is the maximum energy of the electrons emitted when light of $7.3 \times 10^{14} \text{ Hz}$ is used?</p> <p style="text-align: center;">OR</p> <p>A metallic surface, when illuminated with light of wavelength λ_1, emits electrons with energies upto a maximum value E_1, and when illuminated with light of wavelength λ_2, where $\lambda_2 < \lambda_1$, it emits electrons with energies upto a maximum value E_2. Prove that Planck's constant h and the work function ϕ of the metal are given by;</p> $h = \frac{(E_2 - E_1)\lambda_1\lambda_2}{c(\lambda_1 - \lambda_2)} \quad \text{and} \quad \phi = \frac{E_2\lambda_2 - E_1\lambda_1}{(\lambda_1 - \lambda_2)}$		
<p>SECTION-C (2Q × 20 M = 40 Marks)</p>			
Q 10.	<p>(a) Derive an expression to establish a relation between the acceptance angle and the refractive indices of the core and the cladding.</p> <p>(b) Discuss the photoelectric effect with diagram and various characteristic graphs.</p>	<p>10</p> <p>10</p>	<p>CO1</p> <p>CO4</p>
Q 11.	<p>(a) Derive the expression of the pair production and pair annihilation. Justify that the pair production phenomenon cannot happen in empty space.</p> <p>(b) In a certain conducting region, $\mathbf{H} = yz(x^2 + y^2) \mathbf{a}_x - y^2xz \mathbf{a}_y + 4x^2y^2 \mathbf{a}_z \text{ A/m}$. Determine the value of \mathbf{J} at (5, 2, -3).</p> <p style="text-align: center;">OR</p> <p>(a) Derive an expression for a normalized wave function for a particle of mass m moving in a one-dimension box of length L. Use schematic diagrams to analyze the behavior of wave function and probability density of wave function in the box.</p> <p>(b) A steady current element of $10^{-3} \mathbf{a}_z \text{ A.m}$ is located at the origin in free space. Determine the magnetic field intensity due to this current element at (1, 0, 0).</p>	<p>15</p> <p>5</p> <p>15</p> <p>5</p>	<p>CO4</p> <p>CO3</p> <p>CO4</p> <p>CO3</p>

Standard Physics Constants and their values:			
Constants	Standard values		
Planck's constant (h)	$6.626 \times 10^{-34} \text{ Js}$		

	Permittivity of free space (ϵ_0)	$8.854 \times 10^{-12} F/m$			
	Velocity of light (c)	$3 \times 10^8 m/s$			
	Boltzmann constant (k_B)	$1.38 \times 10^{-23} J/K$			
	Rest mass of an electron (m_0)	$9.11 \times 10^{-31} kg$			
	Charge on electron (e)	$1.6 \times 10^{-19} C$			