Enrolment No:



UNIVERSITY OF PETROLEUM AND ENERGY STUDIES End Semester Examination, December 2021

Course: Thermodynamics and Phase Behavior

Program: B.Tech – Petroleum Engineering (Upstream)

Semester: III

Time 03 hrs.

Course Code: MEPD2007 Max. Marks: 100

SECTION A

	Each Question will carry 5 Marks Instruction: Write the statement / answer(s) $5 \times 4 \text{ M} = 20 \text{ M}$	
S. No.	Question	CO
Q 1	Discuss a pure substance. Is iced water a pure substance? Why?	CO1
Q2	What is the difference between an ideal and a perfect gas?	CO1
Q3	Write a short note on (i) Mollier chart (ii) Compressibility Factor	
Q4	What is a perpetual motion machine of second kind? Why is it impossible?	CO2 CO1
Q5	A refrigerator based on reversed Carnot cycle works between two such temperatures that the ratio between the low and high temperature is 0.8. If a heat pump is operated between the same temperature range, then what would be its COP?	CO3
	SECTION B Each Question will carry 10 Marks	
2.	Instruction: Write short / brief notes $4 \times 10 \text{ M} = 40 \text{ M}$	
Q6	A reversible change has quasi-static characteristics, but a quasi-static process may not be reversible one, Justify.	CO1
Q7	A mass of gas is compressed in a quasi-static process from 70 kPa, 0.1 m ³ to 0.4 MPa, 0.03 m ³ . Assuming that the pressure and volume are related by PV ⁿ =constant, find the work done by the gas system.	CO2
Q8	Derive an expression for the ideal efficiency of an Otto cycle.	CO3
Q9	Wet steam at 165 °C and dryness fraction of 0.75 is heated at a constant pressure until it becomes superheated vapour at 300 °C. Find the change in specific volume, enthalpy, and entropy.	CO3

Section C

- 1. Each Question will carry 20 Marks
- 2. Instruction: Write long answer.

 $2 \times 20 \text{ M} = 40 \text{ M}$

A flow rate of 0.42 kg/s is maintained in a steam turbine under steady flow conditions as it receives steam with an enthalpy 3240 kJ/kg, velocity 35 m/s and elevation 4 m. The outlet of steam from the turbine has enthalpy of 2450 kJ/kg, velocity 125 m/s, and elevation 1 m. In the entire process the heat lost takes place at the rate of 0.25 kJ/s. Determine the power output of the turbine in kW?

CO4

Q11 Define entropy. What do you understand by entropy principle? What are the causes of entropy increases?

CO₂

OR

With the help of pressure-volume diagram explain the working of an air standard Diesel cycle and state the assumptions made.

Data for Q9
Saturated Water and Steam (Temperature) Tables

(t)	(p)	(v_f)	(v_g)	(h_f)	(h_{fg})	(h_g)	(s_f)	(s_{fg})	(s_g)	(t)
120	1.985 4	0.001 061	0.891 52	503.7	2 202.3	2 706.0	1.528	5.601	7.129	120
122	2.114 5	0.001 063	0.840 45	512.2	2 196.6	2 708.8	1.549	5.559	7.108	122
124	2.250 4	0.001 064	0.792 83	520.7	2 190.9	2711.6	1.570	5.517	7.087	124
126	2.393 3	0.001 066	0.748 40	529.2	2 185.2	2714.4	1.592	5.475	7.067	126
128	2.543 5	0.001 068	0.706 91	537.8	2 179.4	2717.2	1.613	5.433	7.046	128
130	2.701 3	0.001 070	0.668 14	546.3	2 173.6	2719.9	1.634	5.392	7.026	130
132	2.867 0	0.001 070	0.631 88	554.8	2 167.8	2 722.6	1.655	5.351	7.006	132
134	3.040 7	0.001 072	0.597 95	563.4	2 161.9	2 725.3	1.676	5.310	6.986	134
136	3.222 9	0.001 074	0.566 18	572.0	2 155.9	2 727.9	1.697	5.270	6.967	136
138	3.413 8	0.001 078	0.536 41	580.5	2 150.0	2 730.5	1.718	5.229	6.947	138
140	3.613 9	0.001 080	0.508 49	589.1	2 144.0	2 733.1	1.739	5.189	6.928	140
142	3.823 1	0.001 080	0.482 30	597.7	2 137.9	2 735.6	1.760	5.150	6.910	142
144	4.042 0	0.001 082	0.457 71	606.3	2 131.8	2 738.1	1.780	5.111	6.891	144
146	4.270 9	0.001 084	0.434 60	614.9	2 125.7	2 740.6	1.801	5.071	6.872	146
148	4.510 1	0.001 080	0.412 88	623.5	2 119.5	2 743.0	1.821	5.033	6.854	148
150	4.760 0	0.001 091	0.392 45	632.2	2 113.2	2 745.4	1.842	4.994	6.836	150
155	5.433 3	0.001 091	0.346 44	653.8	2 097.4	2 751.2	1.892	4.899	6.791	155
160	6.180 6	0.001 102	0.306 76	675.5	2 081.2	2 756.7	1.943	4.805	6.748	160
165	7.007 7	0.001 102	0.272 40	697.2	2 064.8	2 762.0	1.992	4.713	6.705	165
170	7.920 2	0.001 114	0.242 55	719.1	2 048.0	2 767.1	2.042	4.621	6.663	170

Specific	Enthalpy	of Su	perheated	Steam
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(p)	(t_s)	100	150	200	250	300	350	400	500	600	700	800
2.5	127.4		2 764.5	2 868.0	2 969.6	3 070.9	3 172.8	3 275.9	3 486.5	3 703.6	3 927.3	4 157.6
3.0	133.5		2 760.4	2 865.5	2 967.9	3 069.7	3 171.9	3 275.2	3 486.0	3 703.2	3 927.0	4 157.
3.5	138.9		2 756.3	2 863.0	2 966.2	3 068.4	3 170.9	3 274.4	3 485.4	3 702.7	3 926.7	4 157.
4.0	143.6		2 752.0	2 860.4	2 964.5	3 067.2	3 170.0	3 273.6	3 484.9	3 702.3	3 926.4	4 156.
4.5	147.9		2 746.7	2 857.8	2 962.8	3 066.0	3 169.1	3 272.9	3 484.3	3 701.9	3 926.1	4 156.
5.0	151.8			2 855.1	2 961.1	3 064.8	3 168.1	3 272.1	3 483.8	3 701.5	3 925.8	4 156.
6.0	158.8			2 849.7	2 957.6	3 062.3	3 166.2	3 270.6	3 482.7	3 700.7	3 925.1	4 155.
7.0	165.0			2 844.2	2 954.0	3 059.8	3 164.3	3 269.0	3 481.6	3 699.9	3 924.5	4 155.
8.0	170.4			2 838.6	2 950.4	3 057.3	3 162.4	3 267.5	3 480.5	3 699.1	3 923.9	4 155.
9.0	175.4			2 832.7	2 946.8	3 054.7	3 160.5	3 266.0	3 479.4	3 698.2	3 923 3	4 154.

Specific Entropy of Superheated Steam

(p)	(t_s)	100	150	200	250	300	350 3	(50. 400	500	600	700	800
2.5	127.4		7.169	7.400	7.604	7.789	7.960	8.119	8.410	8.674	8.917	9.142
3.0	133.5		7.077	7.312	7.518	7.703	7.874	8.034	8.326	8.590	8.833	9.058
3.5	138.5		6.998	7.237	7.444	7.631	7.802	7.962	8.254	8.518	8.761	8.986
4.0	143.6		6.929	7.171	7.380	7.568	7.740	7.899	8.192	8.456	8.699	8.925
4.5	147.9		6.866	7.112	7.323	7.512	7.684	7.844	8.137	8.402	8.645	8.870
5.0	151.8			7.059	7.272	7.461	7.634	7.795	8.088	8.353	8.596	8.821
6.0	158.8			6.966	7.183	7.374	7.548	7.709	8.003	8.268	8.511	8.737
7.0	165.0			6.886	7.107	7.300	7.475	7.636	7.931	8.196	8.440	8.665
8.0	170.4			6.815	7.040	7.235	7.411	7.573	7.868	8.134	8.377	8.603
9.0	175.4			6.751	6.980	7.177	7.354	7.517	7.812	8.079	8.323	8.549

Specific	Volume of	Superheated	Steam
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(p)	(t_s)	100	150	200	250	300	350	400	500	600	700	800
2.5	127.4		0.764 1	0.8620	0.957 4	1.052	1.145	1.239	1.424	1.610	1.795	1.980
3.0	133.5		0.633 7	0.7164	0.7964	0.875 3	0.953 5	1.031	1.187	1.341	1.496	
3.5	138.9		0.540 6	0.6123	0.6814	0.749 3	0.8166	0.883 5	1.017	1.149		1.650
4.0	143.6		0.470 7	0.534 3	0.595 2	0.654 9	0.713 9	0.772 5	0.889 2		1.282	1.414
4.5	147.9		0.416 5	0.473 8		0.5817	0.634 3			1.005	1.121	1.237
4	147.5		0.410 3	0.473 8	0.528 4	0.361 /	0.034 3	0.686 5	0.790 5	0.893 9	0.997 1	1.100
5.0	151.8			0.425 0	0.474 4	0.522 6	0.570 1	0.6172	0.7108	0.804 0	0.006.0	0.000
6.0	158.8			0.3520	0.393 9	0.434 4	0.474 2	0.513 6	0.591 8		0.896 9	0.989
7.0	165.0			0.299 9	0.336 4	0.371 4	0.405 7	0.439 6		0.669 6	0.747 1	0.824
8.0	170.4			0.260 8	0.293 2	0.324 1			0.5069	0.573 7	0.640 2	0.706
			• •	A STATE OF THE PARTY OF THE PAR			0.354 3	0.384 2	0.443 2	0.501 7	0.5600	0.618
9.0	175.4			0.230 3	0.259 6	0.287 4	0.3144	0.341 0	0.393 6	0.445 8	0.497 6	0.549
												0.047