

Name:	
Enrolment No:	

**UNIVERSITY OF PETROLEUM AND ENERGY STUDIES**  
**End Semester Examination, Dec 2019**

**Course:** Thermodynamics and phase behavior (MEPD2005)  
**Program:** APE UP  
**Time:** 3 hrs.

**Semester:** III  
**Max. Marks:** 100

**Instructions:** answer all the questions. Internal choice is given.

**SECTION A**  
**Type the answer**

S. No.		Marks	CO
Q 1	Explain the following terms: (a) Kelvin-Planck statement (b) Clausius statement, (c) Carnot theorem, (d) Clausius inequality and (e) Perpetual motion machine of second kind.	5	CO3
Q 2	How is the concept of entropy and unavailability energy related to each other	5	CO3
Q 3	Discuss the use of air standard cycle analysis for study of the I.C. engine.	5	CO6
<b>Type the answer with correct match (no partial marking) for Q4, 5, and 6</b>			
Q 4	1. In spark ignition four stroke cycle engine cam shaft runs 2. Spark ignition engine 3. Compression ignition system 4. For maximum power of SI engine the fuel air ratio is 5. Ratio of break power to fuel energy input	5	CO6
Q 5	1. A reverse Carnot cycle has COP equal to 4. The ratio of higher temperature to lower temperature 2. Reverse Carnot cycle assume all process 3. A Carnot refrigerator requires 70 KJ/min of work to produce 1 ton of refrigeration at -40 degree centigrade	5	CO2

	4. The COP of Carnot refrigerator is 3 and produce 1 ton of refrigeration. The work done is			
Q 6	<ol style="list-style-type: none"> <li>1. Triple point of water</li> <li>2. The specific volume of ice when heating from zero degree centigrade</li> <li>3. At critical point of water</li> <li>4. Throttling calorimeter used for measuring dryness fraction above</li> </ol>	<ol style="list-style-type: none"> <li>(a) solid vapor and liquid coexist</li> <li>(b) ice have maximum volume</li> <li>(c) increases steadily</li> <li>(d) increases and decreases</li> <li>(e) <math>p = 231.2</math> bar</li> <li>(f) <math>v = 0.00317</math> m<sup>3</sup>/kg</li> <li>(g) 0.98</li> <li>(e) 0.62</li> </ol>	5	CO5

**SECTION B**  
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Q 7	A reversible engine operates between temperatures $T_1$ and $T$ ( $T_1 > T$ ). A second reversible engine at the same temperature “ $T$ ” receives the energy rejected from this engine. The second engine rejects energy at temperature $T_2$ ( $T_2 < T$ ). Show that temperature $T$ is the arithmetic mean of temperatures $T_1$ and $T_2$ if the engines produce the same amount of work output.	10	CO1
Q 8	It is given that temperature of the source and sink are equal to $T_h$ and $T_L$ . If the source and sink are finite i.e. as the heat engine operates the temperature of source fall and temperature of sink rises to an equilibrium temperature $T_f$ . By the entropy principle prove that the $T_f$ is an geometric mean of $T_H$ and $T_L$ .	10	CO4
Q 9	0.2 kg of air at 300°C is heated reversibly at constant pressure to 2066 K. Find the available and unavailable energies of the heat added. Take $T_0 = 30^\circ\text{C}$ and $C_p = 1.0047$ kJ/kg K.	10	CO6
Q10	<p>Prove that for an ideal gas</p> $S_2 - S_1 = C_p \text{Log} \left( \frac{V_2}{V_1} \right) + C_v \text{Log} \left( \frac{P_2}{P_1} \right)$ <p>The terms have their usual meanings as ‘S’ is entropy, ‘P’ is pressure, ‘V’ is volume at thermodynamic state 1 and 2. <math>C_p</math> and <math>C_v</math> are the co-efficient at constant pressure and constant volume.</p>	10	CO2

Q11	<p>Two Carnot engines A and B are connected in series between two thermal reservoirs maintained at 1000 K and 100 K respectively. Engine A receives 1680 kJ of heat from the high-temperature reservoir and rejects heat to the Carnot engine B. Engine B takes in heat rejected by engine A and rejects heat to the low-temperature reservoir. If engines A and B have equal thermal efficiencies, determine</p> <p>(a) The heat rejected by engine B</p> <p>(b) The temperature at which heat is rejected by engine A.</p>	10	CO3
<p><b>SECTION-C</b> <b>Upload pdf file</b></p>			
Q 12.	<p>Two engines are operating on ideal Otto cycle and Diesel cycle for which the following information are available.</p> <p>Maximum temperature = 1277<sup>o</sup> C</p> <p>Exhaust temperature = 477<sup>o</sup> C</p> <p>Ambient condition = 0.1 MPa and 37<sup>o</sup>C</p> <p>Air consumption = 2 kg/min</p> <p>Estimate (a) compression ratio (b) air standard efficiency (c) power output</p> <p style="text-align: center;"><b>OR</b></p> <p>An eight cylinder four stroke petrol engine with bore and stroke of 10 cm each uses volatile fuel of composition C- 84%, H<sub>2</sub>-16 %. The throat diameter of choke tube is 40 mm. The volumetric efficiency at 3000 rpm is 0.75 referred to 0<sup>o</sup> C and 1.013 bar, The pressure depression is 0.116 bar and the temperature at throat is 16<sup>o</sup>C. if chemically correct air fuel ratio is supplied for combustion find (a) fuel consumption in kh/hr and (b) air velocity through tube.</p> <p style="text-align: center;">Assume R for 287 J/Kg <sup>o</sup>K for air and 971 J/Kg <sup>o</sup>K for fuel vapor</p>	20	CO6