

Name:

Enrolment No:



UNIVERSITY OF PETROLEUM AND ENERGY STUDIES

END Semester Examination, December 2019

Programme Name: M.Tech-Energy System

Semester : I

Course Name : Thermodynamics and Heat Transfer Systems

Time : 03 hrs

Course Code : EPEC7028

Max. Marks: 100

Nos. of page(s) : 3

Instructions:

- There are three sections viz. Section A, Section B and Section C. Section A carries 20 marks, Section B carries 40 marks and Section C carries 40 marks
- Attempt all the questions in Section A, B and any two in section C
- Make appropriate assumptions wherever required

SECTION A – 20 Marks

S. No.		Marks	CO
Q 1	<p>A closed thermodynamic system employs the cycle shown in the figure below:</p> <p>If the system performs as a heat engine and the heat transfer to the low temperature heat reservoir is 50 MJ, determine the thermal efficiency of the cycle</p>	5	CO4
Q 2	It is impossible to construct a heat engine working on single reservoir. Interpret?	5	CO4
Q 3	An object is initially at a temperature above that of its surroundings. We have seen that many kinds of convective process will bring the object into equilibrium with its surroundings. Describe the characteristic of a process that will do so with the least net increase of the entropy of the universe.	5	CO4
Q.4	Is it possible to increase the heat transfer from a convectively cooled isothermal sphere by adding insulation? Explain fully.	5	CO1

SECTION-B (40 Marks)

Q 6	Humans are able to control their heat production rate and heat loss rate to maintain a nearly constant core temperature of $T_c = 37^\circ\text{C}$ under a wide range of environmental conditions. This process is called thermoregulation. From perspective of calculating heat transfer between a human body and its surroundings, we focus on a layer of skin	10	CO1
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	<p>and fat, with its outer surface exposed to the environment and its inner surface at a temperature slightly less than the core temperature, $T_i = 35\text{ }^\circ\text{C} = 308\text{ K}$. Consider a person with a skin/fat layer of thickness $L = 3\text{ mm}$ and effective thermal conductivity $k = 0.3\text{ W/mK}$. The person has a surface area $A = 1.8\text{ m}^2$ and is dressed in a bathing suit. The emissivity of skin is $\varepsilon = 0.95$.</p> <p>(a) When the person is in still air at $T_\infty = 297\text{ K}$, What is the skin surface temperature and rate of heat loss to the environment? Convective heat transfer to the air is characterized by a free convection coefficient of $h = 2\text{ W/m}^2\text{K}$.</p> <p>(b) When the person is in water at $T_\infty = 297\text{ K}$, what is the skin surface temperature and heat loss rate? Heat transfer to the water is characterized by a convective coefficient of $h = 200\text{ W/m}^2\text{k}$</p>		
Q 7	<p>Consider a concentric tube heat exchanger with hot and cold-water inlet temperature of 200°C and 35°C respectively. The flow rate of hot and cold fluids is 42 and 84 kg/h, respectively. Assume the overall heat transfer coefficient is $180\text{ W/m}^2\text{K}$. What is the maximum heat transfer rate that could be achieved for the prescribed inlet conditions? If the exchanger is operated in counter flow with heat transfer area of 0.33 m^2. Determine the outlet fluid temperature.</p>	10	CO3
Q.8	<p>Calculate the net radiant heat exchange per m^2 area for two large parallel plates at temperatures of 427°C and 27°C. ε (hot plate) = 0.9 and ε (cold plate) = 0.6. If a polished aluminum shield is placed between them, Compute the % reduction in the heat transfer ε (shield) = 0.4</p> <div style="text-align: center;"> </div>	10	CO1
Q.9	<p>A heat engine receives heat from a source at 1500 K at a rate of 700 kJ/s, and it rejects the waste heat to a medium at 320 K. The measured power output of the heat engine is 320 kW, and the environment temperature is 25°C. Determine (a) the reversible power, (b) the rate of irreversibility, and (c) the second-law efficiency of this heat engine.</p>	10	CO5
SECTION C (40 Marks)- Attempt any two			
Q 10	<p>At a particular instant of time, a square metal bar has an axial temperature distribution given by: $T(x) = 50(1 + 8x^2)$ where x is the distance (in meters) measured from one end and T is the local temperature (in $^\circ\text{C}$). Due to its high thermal conductivity, the temperature in the bar may be assumed uniform at any cross-section. The cross-section of the bar has width $W = 2.5\text{ cm}$ and the length of the bar is $L = 0.3\text{ m}$. The density and specific heat of the metal are $\rho = 2700\text{ kg/m}^3$ and $c = 0.90\text{ J/kg-K}$, respectively.</p> <p>a.) Is the average bar temperature rising or falling at this instant of time? (Assume that the bar can only transfer energy at its end points; i.e., the sides are insulated.)</p>	20	CO1

	<p>b.) Calculate the change in internal energy if the bar is cooled to a uniform temperature of $T_f = 20^\circ\text{C}$.</p> <p>c.) Calculate the change in entropy of the bar for the process in part (b).</p> <p>d.) What is the change in exergy of the bar for the process in part (b) given a large heat sink at 20°C?</p> <p>e.) What is the maximum thermal efficiency at which work could be produced for the conditions in part (d)?</p>		
Q. 11	<p>The sketch below shows an ideal experiment done in a perfectly insulated, rigid container with compartments separated by a frictionless piston. The two compartments contain different amounts of the same gas ($m_A = 1.2 m_B$). The piston is nonadiabatic (heat can be transferred) and moves very slowly. Compartment A is initially at a higher temperature than compartment B ($T_A > T_B$) but the pressure in the two compartments is the same. Assume an ideal gas with constant c_p and c_v.</p> <div style="display: flex; align-items: center; margin: 10px 0;"> <div style="border: 1px solid black; padding: 5px; display: flex; justify-content: space-around; width: 150px; height: 50px;"> <div style="text-align: center; width: 45%;">A</div> <div style="width: 10%; border-left: 2px solid black;"></div> <div style="text-align: center; width: 45%;">B</div> </div> <div style="margin-left: 10px;"> <p>Initially: $P_A = P_B$ $T_A > T_B$</p> </div> </div> <p>a) Which way, if at all, does the piston move? Indicate by an arrow on the drawing and give an explanation (5 Marks).</p> <p>b) Evaluate final temperatures and volumes in compartments A and B? Express your answer in terms of given properties (10 Marks).</p> <p>c) Evaluate net work done in the experiment (5 Marks)?</p>	20	CO4
Q.12	<p>The side of a building of height $H = 7\text{ m}$ and length $W = 30\text{ m}$ is made entirely of glass. Estimate the heat loss through this glass (Ignore the thermal resistance of the glass) when the temperature of the air inside the building is 20°C, the outside air temperature is -15°C and a wind of 15 m/s blows parallel to the side of the building. Select the appropriate correlation from those listed below of local Nusselt number to estimate the average heat transfer coefficient. For air take: $\rho = 1.2\text{ kg/m}^3$, $\mu = 1.8 \times 10^{-5}\text{ kg/m s}$, $C_p = 1\text{ kJ/kg K}$ and $Pr = 0.7$</p> <ul style="list-style-type: none"> • Free convection in air, laminar ($Gr_x < 10^9$): $Nu = 0.3 Gr_x^{1/4}$ • Free convection in air, turbulent ($Gr_x > 10^9$): $Nu = 0.09 Gr_x^{1/3}$ • Forced convection, laminar ($Re_x < 10^5$): $Nu = 0.33 Re_x^{0.5} Pr^{1/3}$ • Forced convection, turbulent ($Re_x > 10^5$): $Nu = 0.029 Re_x^{0.8} Pr^{1/3}$ 	20	CO2